hope of obtaining chemical "peptidases" more selective than enzymes.

(8) Visiting Scientist at the National Institutes of Health on leave of absence from the Weizmann Institute, Rehovoth, Israel.

NATIONAL INSTITUTE OF ARTHRITIS

AND METABOLIC DISEASES ABRAHAM PATCHORNIK⁸ NATIONAL INSTITUTES OF HEALTH WILLIAM B. LAWSON BETHESDA 14, Md. BERNHARD WITKOP RECEIVED JUNE 23, 1958

MICROBIOLOGICAL TRANSFORMATION OF RAUWOLFIA ALKALOIDS

Sir:

In recent years the study of the actions of microorganisms on steroids has led to the discovery of many new transformations difficult to carry out by chemical means, and to production of new important pharmacologically active compounds. A study of the action of microörganisms on alkaloids particularly of the indole type, appeared especially attractive since the susceptibility of the indole moiety to attack by many chemical oxidative processes limits changes that may be made in other parts of the molecule.¹ We now wish to report the 18α -hydroxylation of yohimbine and α -yohimbine by *Streptomyces aureofaciens* (ATCC 11834) and *Streptomyces rimosus* (NRRL 2234).

The Streptomyces cultures were grown in a medium containing 1.5% soybean meal, 2.5%glucose, 0.25% calcium carbonate and 0.5 mg./ml. of yohimbine hydrochloride. After incubation at 25° on a rotary shaker set at 280 r.p.m. with a 2-inch stroke for 1-2 weeks, the conversion could be demonstrated by subjecting the fraction extractable with chloroform at $\bar{p}H$ 9-10 to paper chromatographic analysis using the solvent system *i*-amyl alcohol-carbon tetrachloride-propionic acid (75:60:2) against water vapor equilibrated Whatman No. 1 paper.² The product appeared as a spot detectable by fluorescence, ultraviolet absorption and ferric ferricyanide spray, with an R_t value of 0.15 (R_t of yohimbine, 0.50). A control fermentation with no added yohimbine and an uninoculated medium containing yohimbine did not give rise to this product. By using the same extraction and chromatographic procedure, the chloroform extractable material from 4.5 l. of fermentation broth (45 flasks) was separated on twelve sheets of paper 10.5 inches wide. The appropriate band was eluted with methanol, and the eluate was crystallized from ethyl acetate-acetone to give 117 mg. of colorless cubes, m.p. 252-252.5°, $[\alpha]D+37°$ (methanol. The new substance was assigned structure Ib on the basis of the evidence: analysis gave the composition, $C_{21}H_{26}O_4N_2$ (found: C, 67.93; H, 7.07; CH₃O, 8.7; eq. wt. (perchloric acid), 367), corresponding to the addition of one oxygen atom to yohimbine (Ia). That the oxygen was present as a secondary hydroxyl group was shown by the formation of a diacetate (Ic) (m. p. $307-307.5^{\circ}$, $[\alpha]p-39^{\circ}$ (chloroform); found: C, 66.44; H, 6.72; CH₃CO, 19.7) on treatment with pyridine

(1) Since completion of this work W. O. Godtfresen, et al., Experientia, 14, 88 (1958), have reported the microbiological hydroxylation of apoyohimbine, 3-epiapoyohimbine and β -yohimbine methyl ether.

(2) W. T. Sokolski, S. Ullman, H. Koffler and P. A. Tetrault, Antibiotics and Chemotherapy, 4, 1057 (1957). and acetic anhydride. The substance gave an acetonide (II) (m. p. 258-259°, found : C, 70.40; H, 7.43) by reaction with acetone-perchloric acid, indicating that the new hydroxyl group was situated at C-18 and *cis* to the C-17 hydroxyl function. The presence of a 1,2-glycol system was evidenced by the formation of a bis-2,4-dinitrophenylhydrazone derivative after treatment of Ib with periodic acid.



When the same fermentation procedure was applied using α -yohimbine³ (IIIa) as substrate, a new substance moving with an R_t of 0.17 (R_t of α -yohimbine 0.45) was detected. The product was isolated from the fermentation in the manner described above for 18α -hydroxyyohimbine. The hydrochloride crystallized in colorless needles from methanol-hydrochloric acid, m. p. 288-290° (C_{21} - $H_{26}O_4N_2$. HCl. 0.5H₂O, found: C, 60.39; H, 7.00). It was shown to be 18α -hydroxy- α -yohimbine (IIIb) since it formed a diacetate (IIIc), m.p. 278-279°, ($[\alpha]$ p-14.5° (chf.), found C, 65.86; H, 6.58; CH₃CO, 19.47), an acetonide, m. p. 144-146°, and reacted with periodic acid to give a *bis*-2,4-dinitrophenylhydrazone.

(3) A. LeHir, M. M. Janot and R. Goutarel, Bull. soc. chim. France, 20, 1027 (1953).

THE SQUIBE INSTITUTE FOR MEDICAL RESEARCH New Brunswick, New Jersey

RECEIVED AUGUST 1, 1958

STRUCTURE OF A NEW ANTIBIOTIC, PYOLUTEORIN Sir:

A new antibiotic, pyoluteorin, has been isolated from cultures of *Pseudomonas aeruginosa*, T 359 and IFO 3455,¹ and shown to have a powerful antibacterial activity *in vitro*.²

As was previously described,² pyoluteorin (I) $(C_{11}H_7O_8NCl_2)$ has these physical properties: m.p. 174-175°(dec.); $\lambda_{max}^{\text{BioH}}$ 255 m μ (ϵ 4,200), 310 (ϵ 13,000); $\lambda_{max}^{\text{Miol}}$ 3.02 μ (OH,NH), 6.14(conj. C=O); found: C, 48.48; H, 2.86; N, 5.11; Cl, 25.82; mol.wt. 268 (Rast method), no C-CH₃, no N-CH₃,

⁽¹⁾ Inst. for Fermentation, Osaka, List of Cultures, 108 (1956).

⁽²⁾ R. Takeda, J. Fermentation Technology, Osaka, in press.